

7 Reasons Why Sodium Bisulfate Is a Better Choice Than Hydrochloric Acid

And why it's the safer way to lower pH and alkalinity in pools

	<p>No Fumes.</p> <p>Sodium Bisulfate won't fume the way hydrochloric acid does. So there's less danger to eyes, lungs and skin.</p>	<p>No Dangerous Splashes.</p> <p>Sodium Bisulfate is powdered, so it's less reactive if it spills. Since it can be removed easily from the skin with ordinary water, Sodium Bisulfate poses less risk of injury. If you should happen to spill Sodium Bisulfate, just sweep it up and return it to its container.</p>	<p>Fewer Regulations.</p> <p>Unlike hydrochloric acid, Sodium Bisulfate is classified as a non-hazardous material. It requires less local, state and federal regulatory control than other pH-lowering chemicals.</p>
<p>No Special Application Equipment Required.</p> <p>Transporting, storing and applying Sodium Bisulfate is easy. Just keep it in a covered container and apply as directed.</p>	<p>Less Damage to Poolside Accessories.</p> <p>Hydrochloric acid splashes can mar carpeting and furniture. But Sodium Bisulfate can be swept up and returned to its container without damage.</p>	<p>Less Damage to Vehicles and Equipment.</p> <p>Since Sodium Bisulfate doesn't fume, it won't corrode metals during transport and application.</p>	<p>Economical To Use.</p> <p>Not only is Sodium Bisulfate affordable to use, its safer usage characteristics mean fewer unanticipated costs due to spills and fuming.</p>

Adjust pool water the smart way, with Sodium Bisulfate.